## Revision checklist

#### **SP6 Radioactivity**

#### **SP6a Atomic models**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
7 <sup>th</sup>	Describe the structure of an atom (in terms of nucleus and electrons).			
7 <sup>th</sup>	State where most of the mass of an atom is found.			
7 <sup>th</sup>	State the sizes of atoms and small molecules.			
8 <sup>th</sup>	Describe an early model of the atom.			
8 <sup>th</sup>	Describe how and why our model of the atom has changed over time, including the plum pudding model and the Rutherford alpha particle scattering.			

#### SP6b Inside atoms

Step	Learning outcome	Had a look	Nearly there	Nailed it!
7 <sup>th</sup>	State what is meant by an isotope.			
8 <sup>th</sup>	Represent isotopes using symbols.			
8 <sup>th</sup>	Explain how atoms of different elements are different (in terms of numbers of electrons and protons).			
7 <sup>th</sup>	Recall the charges and relative masses of the three subatomic particles.			
8 <sup>th</sup>	Explain why all atoms have no overall charge.			

#### **SP6c Electrons and orbits**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
<b>7</b> <sup>th</sup>	Describe where electrons are found inside atoms (in terms of shells).			
8 <sup>th</sup>	Describe when electrons can change orbit.			
<b>7</b> th	Recall what an ion is.			
8 <sup>th</sup>	Describe how ionisation occurs.			
8 th	Describe some of the evidence for the Bohr model of the atom.			

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#### **SP6d Background radiation**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
9 <sup>th</sup>	Explain what background radiation is.			
9 <sup>th</sup>	Describe how radiation measurements need to be corrected for background radiation.			
8 <sup>th</sup>	List some sources of background radiation.			
8 <sup>th</sup>	Describe how photographic film can be used to detect radioactivity.			
9 <sup>th</sup>	Describe how a Geiger-Müller tube works.			
8 <sup>th</sup>	Describe how the amount of radioactivity can be measured (in terms of the darkness of photographic film or by attaching a counter to a GM tube).			

#### **SP6e Types of radiation**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
7 <sup>th</sup>	Recall the relative masses and relative electric charges of protons, neutrons, electrons and positrons.			
8 <sup>th</sup>	List five types of radiation that are emitted in random processes from unstable nuclei.			
8 <sup>th</sup>	State that the five types of radiation are ionising radiations.			
8 <sup>th</sup>	Describe what alpha and beta particles are.			
8 <sup>th</sup>	Describe the nature of gamma radiation.			
th	Compare the penetrating abilities of alpha, beta and gamma radiation.			
th	Compare the ionisation abilities of alpha, beta and gamma radiation.			

#### **SP6f Radioactive decay**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
9 <sup>th</sup>	Describe the process of β- decay.			
9 <sup>th</sup>	Describe the process of β+ decay.			
10 <sup>th</sup>	Explain how the proton and mass numbers are affected by different kinds of radioactive decay.			
9th	Describe what happens during nuclear rearrangement after radioactive decay.			
10 <sup>th</sup>	Balance nuclear equations for mass and charge.			

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#### SP6g Half-life

Step	Learning outcome	Had a look	Nearly there	Nailed it!
8 <sup>th</sup>	Describe how the activity of a substance changes over time.			
8 <sup>th</sup>	State how half-life can be used to describe the changing activity of a substance.			
8 <sup>th</sup>	Recall the unit of activity.			
8 <sup>th</sup>	Describe how half-life can be used to work out how much of a substance will decay in a certain time.			
10 <sup>th</sup>	Carry out calculations involving half-life.			

#### SP6h Using radioactivity

Step	Learning outcome	Had a look	Nearly there	Nailed it!
7 <sup>th</sup>	Describe how radioactivity is used in smoke alarms.			
7 <sup>th</sup>	Describe how radioactivity is used in irradiating food.			
7 <sup>th</sup>	Describe how radioactivity is used in sterilising equipment.			
7 <sup>th</sup>	Describe how radioactivity is used in tracing and thickness gauging.			
6 th	Recall that radioactivity is used in cancer diagnosis and treatment.			

#### SP6i Dangers of radioactivity

Step	Learning outcome	Had a look	Nearly there	Nailed it!
8 <sup>th</sup>	Describe the hazards of ionising radiation in terms of tissue damage and possible mutations.			
9th	Explain how the dangers of ionising radiation depend on the half-life.			
114	Explain the precautions taken to reduce the risks from radiation and ensure the safety of patients exposed to radiation, and link these to the half-lives of the sources used.			
9 <sup>th</sup>	Explain the precautions taken to reduce the risks from radiation and protect people who work with radiation.			
9th	Describe the differences between contamination and irradiation effects.			
Ith	Compare the hazards of contamination and irradiation.			

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#### SP6j Radioactivity in medicine

Step	Learning outcome	Had a look	Nearly there	Nailed it!
9 th	Describe the advantages and disadvantages of treating tumours with radiation applied internally.			
9 th	Describe the advantages and disadvantages of treating tumours with radiation applied externally.			
8 <sup>th</sup>	Explain the use of radioactive tracers in diagnosis.			
8 <sup>th</sup>	Explain the use of PET scanners in diagnosis.			
8 <sup>th</sup>	Explain why isotopes used in PET scanners have to be produced nearby.			

#### SP6k Nuclear energy

Step	Learning outcome	Had a look	Nearly there	Nailed it!
6 th	Describe some advantages of using nuclear power to generate electricity.			
6 th	Describe some disadvantages of using nuclear power to generate electricity.			
10 <sup>th</sup>	Evaluate the use of nuclear power to generate electricity.			
7 <sup>th</sup>	Describe three types of nuclear reaction.			
6 th	Recall that nuclear reactions can be a source of energy.			

#### **SP6I Nuclear fission**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
8 <sup>th</sup>	Describe the products of the fission of U-235.			
7 <sup>th</sup>	Describe what a chain reaction is.			
8 <sup>th</sup>	Explain how a chain reaction is controlled in a nuclear power station.			
7 <sup>th</sup>	Describe how the thermal energy from a chain reaction is converted to electrical energy.			
7 <sup>th</sup>	Recall that the products of nuclear fission are radioactive.			

#### **SP6m Nuclear fusion**

Step	Learning outcome	Had a look	Nearly there	Nailed it!
7 <sup>th</sup>	Describe what happens in nuclear fusion.			
7 <sup>th</sup>	Recall that nuclear fusion is the energy source for stars.			

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8 <sup>th</sup>	Explain the difference between nuclear fusion and nuclear fission.		
9 <sup>th</sup>	Explain why high temperatures and pressures are needed to make fusion happen.		
10 <sup>th</sup>	Relate the conditions of fusion to the difficulty of making a practical and economic fusion power station.		